
3. (a) Find the ground state energy of a linear harmonic oscillator applying Heisenberg Uncertainty Principle.
(b) Why is the Compton shift not visible in ordinary light?
(c) Define quantum yield of a photoelectric phenomenon.
4. Non-normalized wavefunction of a particle in one dimensional infinitely deep potential well of length $L$ is $\psi(x)=A \sin \left(\frac{n \pi x}{L}\right)$.
(a) Normalize it and check the orthogonality of the wave function.
(b) Find the most probable positions of the particle for $n=1$ and $n=2$ states. $1+1$
5. (a) If U-236 nucleus is fissioned by a neutron, two fission fragments of mass numbers 96 and 138, and two neutrons are obtained. If the masses of the nuclei and neutron are $235.1175,95.9385,137.9487$ and 1.00898 amu , calculate the amount of energy released.
(b) An X-ray photon of wavelength $0.1 \mathrm{~A}^{\circ}$ is reflected at an angle $90^{\circ}$ with its original direction after collision with an electron at rest. Find the energy it loses on collision. $3+2=5$
6. (a) What do you mean by Einstein's A, B coefficients? Show that $\frac{A_{n m}}{B_{n m}}=8 \pi h v^{3} / c^{3}$, where the symbols have their usual meanings.
(b) The spot size of a gas laser beam is 0.5 mm and the wavelength is $6330 \AA$. If the laser cavity is of confocal type, calculate its length.

## Group - B

Answer any two questions:
7. (a) What do you mean by "photoelectric effect"? Ultraviolet light of wavelength 350 nm and intensity $1.00 \mathrm{~W} / \mathrm{m}^{2}$ is directed at a potassium surface. Find the maximum KE of the photoelectrons.
(b) What is radioactivity? Define the decay constant $\lambda$ of a radioactive material. Hence obtain an expression for the number of radioactive atoms at time $t$, given that their initial number was $\mathrm{N}_{0}$. Define half-life of a radioactive element.

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(2+2)+(1+1+3+1)=10
$$

8. (a) Explain the need for a wave equation to describe the behavior of a quantum system. Discuss the required characteristics of such a wave equation.
(b) Starting from the de Broglie's hypothesis, set up the one dimensional Schrodinger wave equation for a free particle. How it is modified if the particle is under the influence of a potential field?
(c) Determine the probability density and the probability current density for a wave function given by (at $t=0) \psi(x)=A \exp \left(-\sigma^{2} x^{2} / 2\right) \exp (i k x)$.

$$
(1+2)+(3+1)+3=10
$$

9. (a) Verify Heisenberg Uncertainty Principle through gamma-ray thought experiment. Can this experiment be done with visible light? Expain. $4+2$
(b) Write de Broglie's relation. Distinguish between macroscopic regime where classical Newton's law of motion is applied and microscopic regime where quantum mechanics to be applied in terms of de Broglie wavelength and dimension of the moving particle.
(c) Establish Bohr's angular momentum quantization condition using de Broglie's relation.
10. (a) Explain the non-existence of electron in a nucleus from Heisenberg's Uncertainty Principle. (H.U.P.)
(b) If the expectation value of the momentum is $\langle p\rangle$ for the wavefunction $\psi(x)$, then what will be the expectation value of momentum for the wavefunction $e^{i k x / \hbar} \psi(x)$.
(c) Evaluate the commutation relations $\left[\hat{x}, \hat{p}_{x}\right]$ and $\left[\hat{x}, \hat{p}_{y}\right]$. Hence write the pair of components of position-momentum commutation relations in terms of Kronecker delta. Have you got any connection of these results with H.U.P.?
(d) Deduce expression for probability current density. Give its physical interpretation.
