

বিদ্যাসাগর বিশ্ববিদ্যালয়

VIDYASAGAR UNIVERSITY

Question Paper

B.Sc. Honours Examinations 2020

(Under CBCS Pattern)

Semester - V

Subject: PHYSICS

Paper: C11T & C11P (Quantum Mechanics and Applications)

> Full Marks : 60 Time : 3 Hours

Candiates are required to give their answer in their own words as far as practicable. The figures in the margin indicate full marks.

Group - A

THEORY (Marks : 40)

Answer any *two* from the following questions : 2×20

- 1. Answer any *five* questions from the following : $5 \times 4=20$
 - (a) What is Gyromagnetic ratio for the orbital and spin motion of an electron in an atom.
 - (b) State Paul's exclusion principle. Give electronic configuration for the element. Ni (Z = 28).

- (c) Find the precessional frequency of an electron orbit when placed in magnetic field of 6 Wb/m².
- (d) Calculate the value of lowest energy of an electron in a one dimensional force free region of length 4A°.
- (e) Distinguish between a classical and quantum harmonic oscillator.
- (f) Calculate the normalization constant for a wave function (at t = 0) given by $\Psi(x) = ae^{-a^2x^2/2}e^{ikx}$ known as a Gaussian wave packet.
- (g) What is Larmor's theorem ?
- 2. (a) (i) Using Heisenberg uncertainty principle, calculate the ground state energy and radius of hydrogen and helium atom.
 - (ii) Using the energy-time uncertaity principle calculate the width of the spectral line when the atom de-exites to the gound state.
 - (iii) Using the uncertainty principle, show that an alpha particle can exist inside a nucleus. 5+2+3
 - (b) (i) State and prove Ehrenfest's theorem.
 - (ii) What are the continuity and boundary conditions that must be satisfied for a wave function to be physically acceptable ?
 - (iii) Show that the uncertainty relation $\Delta x \Delta p \ge \hbar/2$ is satisfied in the case of particle in a one-dimensional box. $\left(\hbar = \frac{\hbar}{2\pi}\right)$
 - (iv) Calculate the probability that a particle in a one-dimensional box a length L can be found between 0.4 L to 0.6 L for the (a) ground state, (b) first excited state, (c) second excited state.
- 3. (a) (i) What is Anomalous Zeeman effect ? Explain with the help of diagram the transition between 3d and 2p levels in a Normal Zeeman effect. 3+2
 - (ii) Calculate the Lande g factor for the following states : (i) $3^{2}S_{1/2}$ (ii) $4^{2}P_{1/2}$.

2+2

(b) (i) The normalized ground state wave function of hydrogen atom is given by,

$$\Psi(r,\theta,\varphi) = \frac{1}{\sqrt{\pi a^{3/2}}} e^{-2r/a}, \ 0 < r < \infty.$$
 Find the value of $\left\langle \frac{1}{r} \right\rangle$

- (ii) Write down the Hamiltonian for the hydrogen atom.
- (iii) Prove that for the hydrogen atom the wave functions Ψ_{100} and Ψ_{200} are orthogonal.
- (iv) What do you mean by the degeneracy of state ? 4+2+2+2
- 4. (a) Define the creation (a^+) and annihilation (a) operators for a harmonic oscillator and show that $\widehat{H} = \hbar \omega \left(a^+ a + \frac{1}{2} \right)$ 5
 - (b) A particle in the harmonic oscillator potential starts out in the state

$$\psi(x,0) = A \left[3\psi_0(x) + 4\psi_1(x) \right]$$

(A) Find A (B) Construct $\psi(x,t)$ and $|\psi(x,t)|$

(c) Suppose a particle starts out in a linear combination of just two stationary states :

$$\psi(x,0) = c \mathbf{1} \psi_1(x) + c \mathbf{2} \psi_2(x)$$

- (i) What is the wave function $\psi(x,t)$ at subsequent times ?
- (ii) Is it a stationary state ?
- (iii) Discuss briefly, with theory, the Stren-Gerlach experiment. Justify the use of a bean of silver atoms in the experiment. 2+3+5

Group - B

PRACTICAL (Marks : 20)

Answer any *one* from the following questions : 1×20

- 1. Demonstrate in details the tunneling effect in tunnel diode using I-V characteristics.
- 2. Solve the s-wave radial Schrodinger equation for the vibrations of hydrogen molecule :

$$\frac{d^2 y}{dr^2} = A(r)u(r), A(r) = \frac{2\mu}{h^2} \left[V(r) - E \right]$$

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Where μ is the reduced mass of the two-atom system for the Morse potential

$$V(r) = D(e^{-2\alpha r'} - e^{-\alpha r'}), r' = \frac{r - r_0}{r}$$

Find the lowest vibrational energy (in MeV) of the molecule to an accuracy of three significant digits. Also plot the corresponding wave function.

Take
$$m = 940 \times 10^6 \text{ eV}/C^2$$
, $D = 0.755501 \text{ eV}$, $\alpha = 1.44$, $r_0 = 0.131349\text{ Å}$

3. Solve the s-wave radial Schrodinger equation for a particle of mass m:

$$\frac{d^2 y}{dr^2} = A(r)u(r), A(r) = \frac{2m}{h^2} \left[V(r) - E \right]$$

For the anharmonic oscillator potential.

$$V(r) = \frac{1}{2}kr^2 + \frac{1}{3}br^3$$

for the ground state energy (in MeV) of particle to an accuracy of three significant digits. Also, plot the corresponding wave function.

Choose
$$m = 940 \text{ MeV} / c^2$$
, $k = 100 \text{ MeV} fm^{-2}$, $b = 0, 10, 30 \text{ MeV} fm^{-3}$.

In these units, $c\hbar = 197.3$ MeV fm. The ground state energy I expected to lie between 90 and 110 MeV for all three cases.